

## SYLLABUS

**1. Course title:**

ORGANIC CHEMISTRY

**2. Code:****3. Cycle of study:**

1

**4. ECTS credits:**

7

**5. Type of course:** Mandatory  Elective**6. Prerequisites:****7. Class restrictions:****8. Duration / semester:**

1

3

**9. Weekly contact hours:**

9.1. Lectures:

4

9.2. Seminars:

0

9.3. Laboratory/Practice classes:

3

**10. Faculty:**

Faculty of Technology

**11. Department/study program:**

Food Technology and Environmental Protection Engineering

**12. Lecturer:**

dr. sc. Zahida Ademović, assoc. prof.

**13. Lecturer's e-mail:**

zahida.ademovic@untz.ba

**14. Web site:**

www.tf.untz.ba

**15. Course aims:**

The aim of this course is to enable students to master the basic principles and fundamental knowledge in the field of organic chemistry in order to understand the profession which is studied. The knowledge acquired will be used by students in solving specific problems of chemical and technological disciplines, as well as problems related to the planning and implementation of the management of technological processes.

**16. Learning outcomes:**

After finishing the course, students will be able to identify, analyze and solve problems of varying complexity, individual or in team, using professional literature in the scientific field of organic chemistry, follow further courses that include subjects of the structure, properties, synthesis and reactions of different organic compounds.

**17. Course content:**

1. Structure and bonding in organic chemistry
2. Nomenclature of Organic Compounds
3. Carbon compounds and chemical bonds, the class of compounds, classification of the reaction in organic chemistry
4. Alkanes and cycloalkanes; conformational and geometrical isomerism
5. Stereochemistry
6. Alkenes, dienes, polyenes
7. Alkynes: properties, synthesis, addition reaction
8. Aromatic compounds: properties and reactions, polycyclic aromatic compounds
9. Alkyl halides; nucleophilic substitution, elimination reactions
10. Alcohols, phenols, aryl halides, ethers, thiols; properties and reactions
11. Aldehydes and ketones: nucleophilic addition to the carbonyl group
12. Carboxylic acid and derivatives
13. Amines and related compounds with nitrogen
14. Heterocyclic compounds
15. Alkaloids and steroids

**18. Learning methods:**

Lectures - through interactive lectures introduce students to the basic concepts and principles of structure and properties of various heterocyclic compounds and natural products and through practical examples and problems explain reaction mechanisms.

Experimental - through practical experiments, students will demonstrate the level of acquired knowledge through lectures and acquire skills for practical and scientific and research work.

Consultation - through consultations, students can deepen the knowledge acquired in classes.

**19. Assessment methods:**

Activity - for activity in lectures and exercises students can obtain 0-5 points.

Colloquium: colloquium of experimental work which consists of a theoretical basis and conducted experimental exercises. For the colloquium student can obtain a maximum of 15 points, the minimum number of points that a student must achieve is 5.

Colloquium and completion of the experimental work is a prerequisite for obtaining a signature.

Written assessment during the semester (Test I and II) - Test I and Test II include problem-solving tasks. The maximum number of points on each test is 15th

Final exam - assessment implies the unification of the entire matter handled.

In order to pass the courses, student must achieve a minimum of 51 points, of which a minimum of 25 points on the final exam.

**20. Assessment components:**

Rating exam is based on the total number of points a student has obtained by completing pre-exam requirements and exams, according to the quality of the acquired knowledge and skills and contains a maximum of 100 points. It is determined according to the following scale:

Obligations	Points
Presence in classes	0-5
Experimental work	5-15
Test I	0-15
Test II	0-15
Final Exam	25-50

**21. Required reading list:**

1. K.Peter, C.Volhardt, Neil E.Schore, *Organska hemija*, Data Status, Beograd, 2004.
2. J.E.McMurry, *Organic chemistry*, Int. Ed. Cengage, USA, 2011
3. J.Budimir, R.Kubiček, S.Marić: *Osnovi preparativne organske hemije*, Univerzitet u Tuzli

**22. Web sources:****23. Applicable starting from the academic year:**

2015/2016

**24. Adopted in the Faculty/Academy session:**